INTERNATIONAL UNION OF PURE AND APPLIED CHEMISTRY

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A CLASSIFICATION OF LINEAR SINGLE-STRAND POLYMERS

(Recommendations 1988)

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A classification of linear single-strand polymers (Recommendations 1988)

The polymer literature lacks, at present, a widely applicable, internally consistent, and easily recognizable system for classifying polymers. This situation prevails in spite of the large number of polymers that are already known or will be synthesized in the future. The classification scheme to be presented here facilitates logical indexing, including through the use of computer techniques, information storage and retrieval. At the present time, the scheme is limited to linear single-strand organic and inorganic polymers. It is consistent with previous IUPAC recommendations (References 1–5).

GENERAL PRINCIPLES

The classification consists of a hierarchical scheme for naming polymers according to the chemical constitution of the repeating units in the main chain (backbone). It can be applied to homopolymers, alternating copolymers, and other macromolecular substances in which the constitutional repeating units can be identified. All existing linear single-strand polymers are embraced by this classification, which has been designed so as to be capable of extension to include any new structures of this type. In view of the open-ended nature of the classification, the tables below can only be illustrative, not comprehensive.

Four hierarchical levels are used; in order of decreasing importance, they are: classes, subclasses, groups, and individual polymers.

CLASSES

Polymers are divided into two principal classes on the basis of the constitution of the main chain.

Homochain polymers are those in which the main chains are constructed from atoms of a single element.

Rule 1

Homochain polymers are named by placing the name or symbol of the element in the main chain immediately before the expression '-chain polymer'.

Examples (see also SUBCLASSES, below, and Table 1) carbon-chain polymer or C-chain polymer; sulfur-chain polymer or S-chain polymer.

Heterochain polymers are those in which the main chains are constructed from atoms of two or more elements.

Class homochain polymers Subclasses carbon-chain polymers silicon-chain polymers boron-chain polymers germanium-chain polymers sulfur-chain polymers phosphorus-chain polymers

Table 1. Examples of subclasses of homochain polymers

Rule 2.1

Heterochain polymers are named by placing the names or symbols of all the elements in the main chain, in parentheses, immediately before the expression '-chain polymer'.

Examples (see also SUBCLASSES, below, and Table 2) (oxygen,carbon)-chain polymer or (O,C)-chain polymer; (oxygen,nitrogen,carbon)-chain polymer or (O,N,C)-chain polymer.

Rule 2.2

The order of citation of the elements in heterochain polymers is that conventionally used in inorganic nomenclature (Reference 6). For the common elements, the order of citation is as follows: O, S, N, P, C, Si, B. Note that in those cases where the main chain has bonds in common with cyclic structures, *all* atoms in the rings must be considered for classification purposes. Thus I is a homochain polymer, whereas II is a heterochain polymer.

On the other hand, a polymer such as III, in which the main chain has no bonds in common with the cyclic structure, is classified as a homochain polymer.

$$C$$
 CH_2 $CH_$

Rule 3.1

The presence of specific sidegroups, or specific elements in sidegroups, is indicated by placing the name of the sidegroup, or of the element in the sidegroup, immediately before the expression '-sidegroup polymer'.

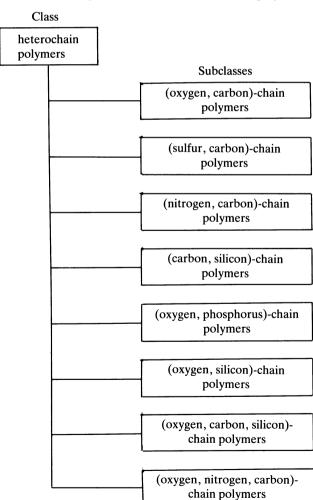


Table 2. Examples of subclasses of heterochain polymers

Rule 3.2

Where more than one element, or more than one sidegroup, is to be specified, the names of the elements or sidegroups are placed in parentheses.

Examples

oxygen-sidegroup polymer; hydroxyl-sidegroup polymer [for poly(1-hydroxyethylene), etc.]; ether-sidegroup polymer [for poly(1-methoxyethylene), etc.]; (O,N,P)-sidegroup polymer; (ether,amide)-sidegroup polymer.

SUBCLASSES

Each class of polymers can be divided into various subclasses, according to the nature of the elements in the main chain. Examples are given in Rules 1, 2.1, as well as in Tables 1 and 2.

GROUPS

Each subclass can be further divided into groups having similar chemical structures. Examples of carbon-chain polymers are polyalkylenes, polyarylenes, etc. (Table 3A).

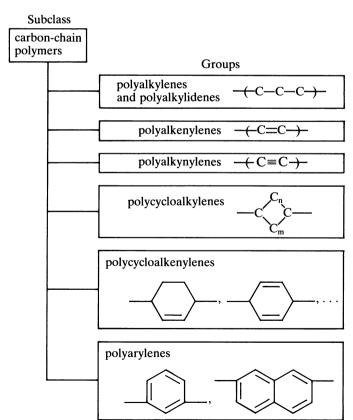


Table 3A. Examples of groups of carbon-chain polymers

Table 3B. Examples of groups of (oxygen,carbon)-chain polymers

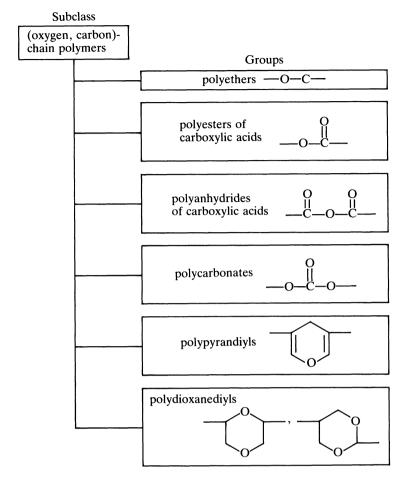


Table 3C. Examples of groups of (sulfur, carbon)-chain polymers

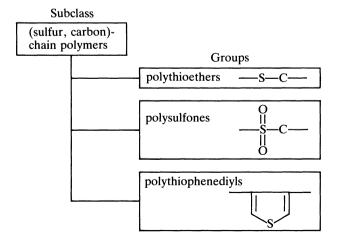
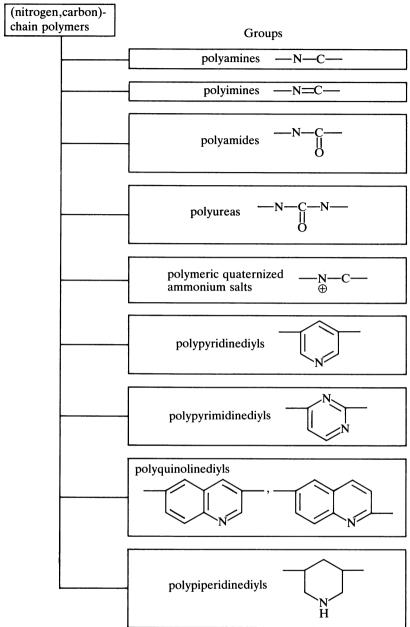


Table 3D. Examples of groups of (nitrogen,carbon)-chain polymers
Subclass

itrogen,carbon)-



Examples of (oxygen,carbon)-chain polymers are polyethers, polycarbonates, etc. (Table 3B). Examples of other types can be found in Tables 3C and 3D.

It is convenient to retain such widely accepted terminology as polyacetals, polycar-bonates, polyamides, polyesters, nucleic acids, etc., for naming the groups of polymers.

INDIVIDUAL POLYMERS

The lowest hierarchical ranking is given to the individual polymers, which are named in accordance with accepted IUPAC nomenclature practice. The reader is referred to References 2 and 3 for recommendations for naming organic and inorganic single-strand polymers, respectively.

It is to be noted that, for polymers of complex structure, assignment to more than one subclass and/or group is sometimes possible. For example, the polymer

may be classified as follows:

class: heterochain

subclass: (O,N,C)-chain polymer

groups: polycarboxylate and/or pyridine-chain polymer

Additional examples are given in Table 4 and the Appendix.

Table 4. Examples of individual polymers

Group		Individual polymers
	(Structure)	(Name)
Polyalkylenes	CH_3 $-(-CCH_2-)_n$ CH_3	Poly(1,1-dimethylethylene) Polyisobutylene
Polyalkenylenes	-(- C=CHCH ₂ CH ₂ -)n Cl	Poly(1-chloro-l-butenylene) Polychloroprene
Polyethers	$ \leftarrow$ O $ \left($ D $\right)_n$	Poly(oxy-1,4-phenylene) Poly(phenylene oxide)
Polyamides	$\begin{array}{c} O & O \\ & \\ \hline NHC(CH_2)_4CNH(CH_2)_6 \\ \hline]_n \end{array}$	Poly(iminoadipoyliminohexamethylene) Poly(hexamethylene adipamide)

REFERENCES

- 1 'Basic Definitions of Terms Relating to Polymers 1974'. Pure Appl. Chem. 40, 477-491 (1974).
- 2 'Nomenclature of Regular Single-Strand Organic Polymers (Rules Approved 1975)'. Pure Appl. Chem. 48, 373-385 (1976).
- 3 'Nomenclature for Regular Single-Strand and Quasi-Single-Strand Inorganic and Coordination Polymers (Recommendations 1984)'. *Pure Appl. Chem.* 57, 149–168 (1985).
- 4 'Stereochemical Definitions and Notations Relating to Polymers (Recommendations 1980)' *Pure Appl. Chem.* **53**, 733–752 (1981).
- 5 'Report on Nomenclature in the Field of Macromolecules'. J. Polym. Sci. 8, 257-277 (1952).
- 6 Rule IP-2.1 in Ref. 3.

APPENDIX

Individu	Individual polymer	Groups	Subclass	Class
1 (S)	catena-poly[sulfur]		S-chain	homochain
2 	poly(1,4-phenylene)	polyarylenes	C-chain	homochain
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	poly(spiro[4.5]dec-2,8-yleneethylene)	polycycloalkylenes	C-chain	homochain
4 (CH ₂ —CH ₂ —),	poly(2,6-biphenylyleneethylene)	polyarylenes	C-chain	homochain
$ \begin{array}{c c} CH_3 \\ C \longrightarrow C \longrightarrow CH_2 \longrightarrow \Pi \end{array} $	poly(1-cyano-1-methylethylene)	polyalkylenes	C-chain	homochain
CI 	catena-poly[dichlorotin]	I	Sn-chain	homochain

Individual polymer	ılymer	Groups	Subclass	Class
$ \begin{array}{ccc} F & CH_3 \\ 7 & \leftarrow Si - Si - h \\ F & CH_3 \end{array} $	catena-poly[(difluorosilicon)(dimethylsilicon)]		Si-chain	homochain
8 (-0-CH ₂ -CH ₂ -0-C-C-) _n	poly(oxyethyleneoxyterephthaloyl)	polyesters of carboxylic acids	(O,C)-chain	heterochain
9 (CH ₂ -) ₁₁	poly[(2-propyl-1,3-dioxane- 4,6-diyl)methylene]	polydioxanediyls, polyacetals	(O,C)-chain	heterochain
10 $-(-0-C-CH-CH_2-C-)_{}$	poly[oxy(2-chlorosuccinyl)]	polyanhydrides of carboxylic acids	(O,C)-chain	heterochain
$\begin{array}{ccc} \operatorname{CH_3} & \operatorname{Br^{\otimes}} \\ & & & & \\ \operatorname{II} & & & & \\ & & & & \\ & & & & \\ & & & & $	poly[(dimethyliminio)- ethylene) bromide]	polymeric quaternized (N,C)-chain ammonium salts	(N,C)-chain	heterochain
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	poly(4,2-pyridinediylimino- 1,4-cyclohexylenebenzylidene)	polyamines and/or polypyridinediyls	(N,C)-chain	heterochain

Indivi	Individual polymer	Groups	Subclass	Class
13 — H — H — CH ₂ — h	poly(4,2-pyridinediyl-4 <i>H</i> -1,2,4-triazole-3,5-diylmethylene)	polypyridinediyls and/ or polytriazolediyls	(N,C)-chain	heterochain
14 N	poly([2,3'-bipyridine]-4,5'-diyl)	polypyridinediyls	(N,C)-chain	heterochain
$\begin{array}{ccc} 0 & & \\ 15 & & \\ & & \\ \end{array}$	poly[imino(1-oxoethylene)]	polyamides (or polypeptides)	(N,C)-chain	heterochain
16 Ph 	catena-poly[(diphenylsilicon)-μ-οχο] polydiphenylsiloxane	polysiloxanes	(O,Si)-chain	heterochain
$\begin{array}{ccc} & H & Ph \\ & & & & \\ & & & & \\ & & & & -(-Al-N) \rightarrow - & & & & & \\ \end{array}$	<i>catena</i> -poly[(hydridoaluminum)-µ- (phenylimido)]	I	(N,Al)-chain	heterochain
18 + Tri-Cri	catena-poly[titaniumtri-µ-chloro]	I	(Cl,Ti)-chain	heterochain
$\begin{array}{c} 19 & -C - NH - C - O - C \\ \parallel & \parallel & \parallel \\ O & O \end{array}$	$-\mathrm{CH_2}$ — $\mathrm{CH_2}$ — $)_{\overline{\mathrm{n}}}$ poly(oxycarbonylimino-1,4-phenyleneiminocarbonyloxyethylene)	nylene- polyurethanes	(O,N,C)-chain	heterochain

Individual polymer	mer	Groups	Subclass	Class
20 (C-O-CH ₂ -)n	poly(3,5-pyridinediylcarbonyloxy-methylene)	polyesters of carboxylic acids and/or polypyridinediyls	(O,N,C)-chain	heterochain
21 + H CH ₂ - H	poly(2,8-phenoxazinediylmethylene)	polyphenoxazine	(O,N,C)-chain	heterochain
22 + O - H - O - CH ₂ O H - O - CH ₂ O H - O - CH ₂ O H - O - O - O - O - O - O - O - O - O	poly(ribocytidylic acid)	nucleic acids; polyesters of phosphoric acid and/or polyfurandiyls	(O,P,C)-chain	heterochain
23 (Ag—NC—) n	$catena ext{-poly}[ext{silver-}\mu ext{-}(ext{cyano-}N\!:\!C)]$	I	(N,C,Ag)-chain	heterochain